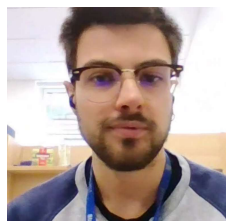




CCP-NC Discussion Meeting 11 May 2023

First-Principles Computation of pNMR Chemical Shifts



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University of St Andrews

History of Chemical-Shift Calculations

closed shell

- | | | | |
|---------|---|---|-----------------------------------|
| 1950 | Theory (Ramsey) | | |
| 1966 | Coupled Hartree-Fock (Lipscomb) | } | Hartree-Fock level |
| 1974 | GIAO (Ditchfield) | | |
| 1983 | IGLO (Kutzelnigg) | | |
| 1990 | SOLO (Bouman, Hansen) | } | including
electron correlation |
| 1992 | GIAO-MP2 (Gauss) | | |
| 1996 | GIAO-CCSD(T) (Gauss) | | |
| 1994 | SOS-DFPT-IGLO (Malkin) | } | DFT
level |
| 1995/96 | GIAO-DFT (Ziegler, Cheeseman, Pulay) | | |
| 1999 | ZORA-DFT (Ziegler) | } | relativistic level |
| 2001 | GIPAW-DFT (Mauri) | } | periodic systems |
| 2003/04 | paramagnetic NMR, $S = \frac{1}{2}$ (Vaara, Patchkovskii) | | |
| 2007/08 | paramagnetic NMR, any S (Kaupp, Vaara) | | |

Theory of pNMR chemical shifts

Milestone papers:

Z. Rinkevicius, J. Vaara, L. Telyatnyk, O. Vahtras, *J. Chem. Phys.* **2003**, *118*, 2550.

S. Moon, S. Patchkovskii, in: *Calculation of NMR and EPR Parameters. Theory and Applications*, M. Kaupp, M. Bühl, V. G. Malkin (Eds.), Wiley-VCH, Weinheim, **2004**, pp. 325-340.

P. Hrobarik, R. Reviakine, A. V. Arbuznikov, O. L. Malkina, V. G. Malkin, F. Koehler, M. Kaupp. *J. Chem. Phys.* **2007**, *126*, 024107.

T. O. Pennannen, J. Vaara, *Phys. Rev. Lett.* **2008**, *100*, 133002

S. A. Rouf, J. Mareš, J. Vaara, *J. Chem. Theory Comput.* **2015**, *11*, 1683.

B. Martin, J. Autschbach, *Phys. Chem. Chem. Phys.* **2016**, *18*, 21051.

G. Mali, M. Mazaj, *J. Phys. Chem. C* **2021**, *125*, 4655.

Theory of pNMR chemical shifts

isotropic magnetic shielding of a nucleus N in a molecule with arbitrary spin multiplicity:

$$\sigma_{\text{iso}} = \underbrace{\sigma_{\text{iso(orb)}}}_{\text{orbital shielding part}} - \underbrace{S(S+1)\beta_e/(3kTg_N\beta_N)[g_e \cdot A_{\text{FC}} + g_e \cdot A_{\text{PC}} + \Delta g_{\text{iso}} \cdot A_{\text{FC}} + 1/3 \text{Tr}(\Delta g_{\text{aniso}} \cdot A_{\text{dip}})]}_{\text{paramagnetic part}}$$

S: total spin

T: temperature

β_e, β_N : Bohr and nuclear magnetons

g_e, g_N : free-electron and nuclear g-values

$\Delta g_{\text{iso}}, \Delta g_{\text{aniso}}$: **g-tensor** elements ($g = g_e + 1 \cdot \Delta g_{\text{iso}} + \Delta g_{\text{aniso}}$)

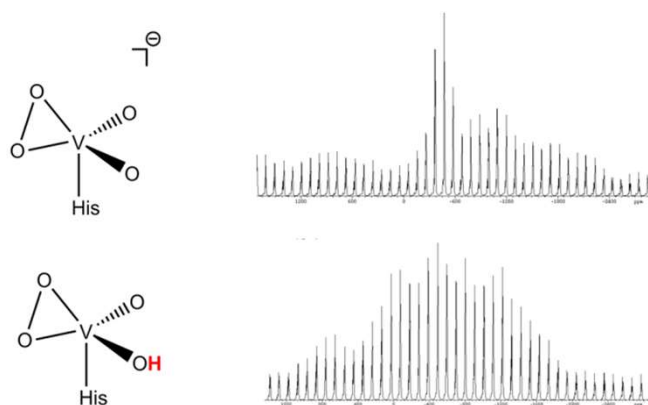
$A_{\text{FC}}, A_{\text{PC}}, A_{\text{dip}}$: **hyperfine coupling tensor** elements

^{13}C and ^1H chemical shifts: $\delta_{\text{iso}} = \sigma_{\text{iso(orb)}}(\text{TMS}) - \sigma_{\text{iso}}$

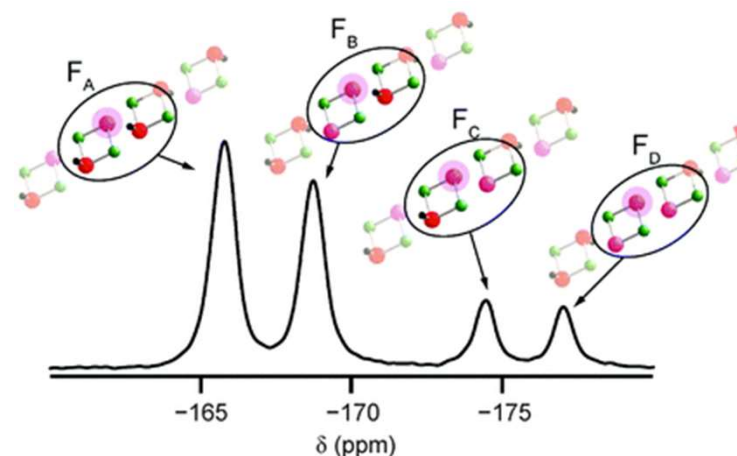
Application of NMR chemical shift computation

- ➔ support for signal assignment
- ➔ information on (local) structure

well established for diamagnetic molecules and materials



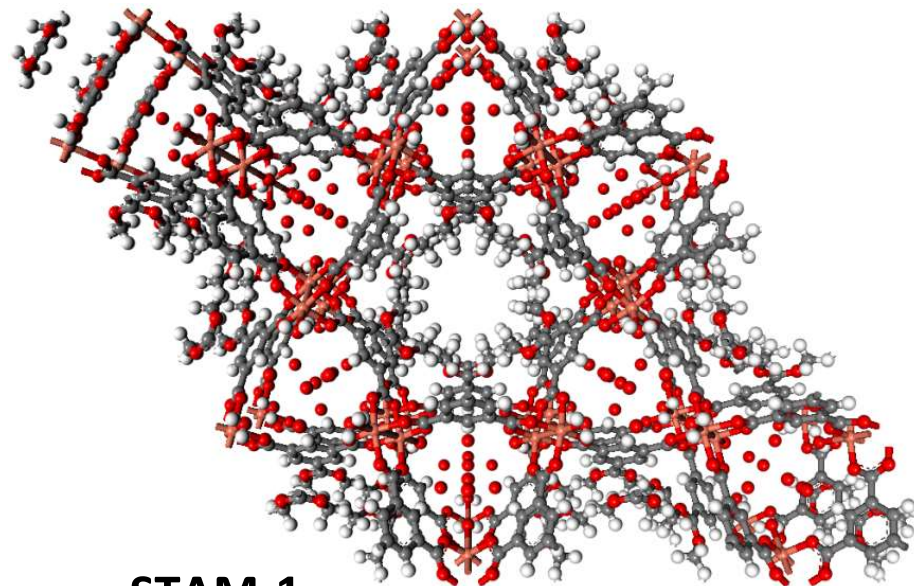
M. Bühl, T. van Mourik, *Wiley Interdisc. Reviews: Comp. Mol. Sci.* **2011**, *1*, 634.



S. E. Ashbrook, D. McKay, *Chem. Commun.* **2016**, *52*, 7186

Metal Organic Frameworks (MOFs)

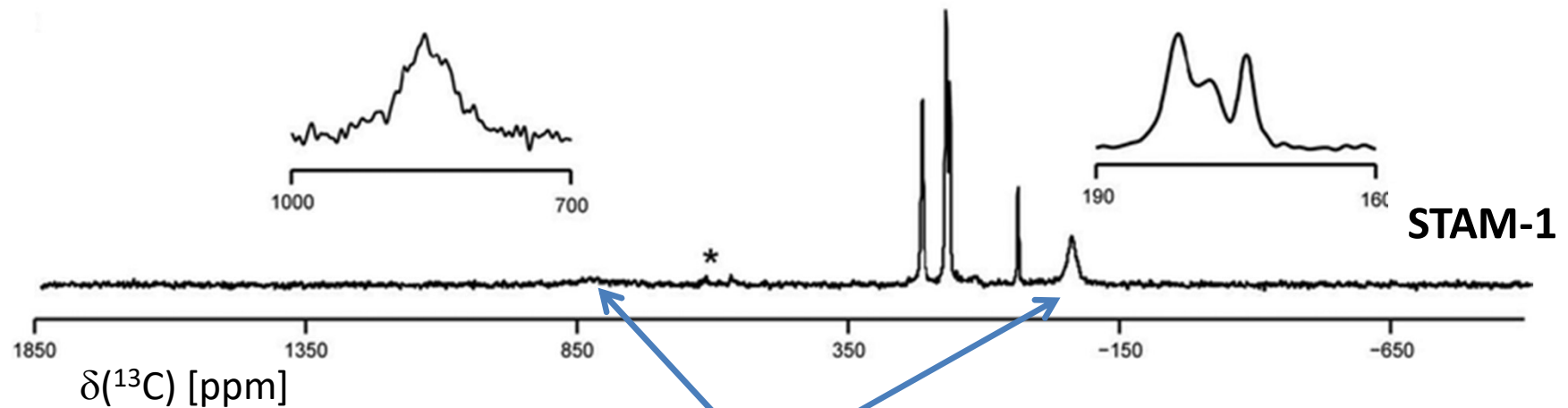
- Gas storage and separation
- Catalysis
- Carrier in medical drug delivery



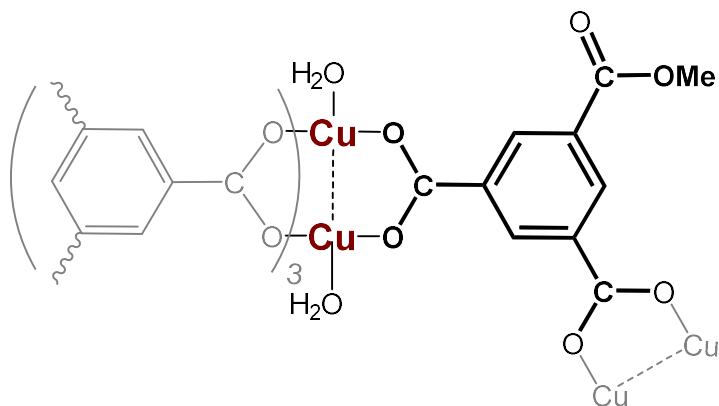
STAM-1

M. I. H. Mohideen, B. Xiao, P. S. Wheatley, A. C. McKinlay, Y. Li, A. M. Z. Slawin, D. W. Aldous, N. F. Cessford, T. Düren, X. Zhao, R. Gill, K. M. Thomas, J. M. Griffin, S. E. Ashbrook, R. E. Morris, *Nat. Chem.* **2011**, 3, 304.

Solid-state NMR of MOFs



huge paramagnetic shifts due to **Cu(II)** centres



signals assigned through
 ^{13}C labelling studies

D. M. Dawson, L. E. Jamieson, M. I. H. Mohideen, A. C. McKinlay, I. A. Smellie, R. Cadou, N. S. Keddle, R. E. Morris, S. E. Ashbrook *Phys. Chem. Chem. Phys.* **2013**, *15*, 919.

Methodology

- Density Functional Theory (DFT)
- Optimisation:
PBE0-D3; basis set: 6-31G*/6-31G**(H^{br})/AE1(Cu) (8s7p4d)
- Orbital shielding, g- and A-tensors:
PBE, PBE0, PBE0-1/3; basis set: IGLO-II(or III)/AE1(Cu) (9s7p4d)

opt, $\sigma_{\text{iso}(\text{orb})}$: Gaussian 09 

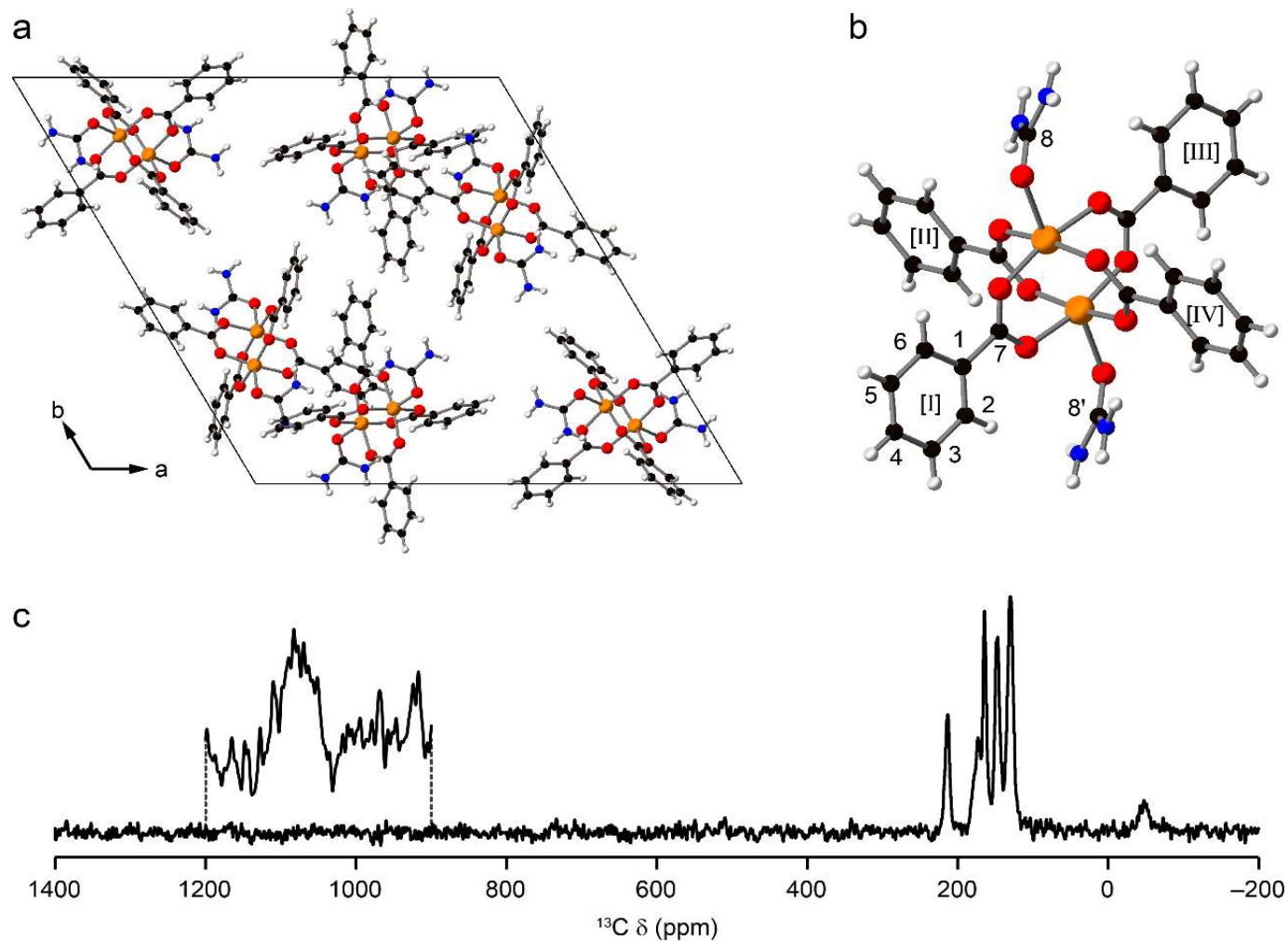
g, A tensors: ORCA 

Validated for mononuclear Cu species (phenolic oximes)

M. Bühl, S. E. Ashbrook, D. M. Dawson, R. A. Doyle, P. Hrobarik, M. Kaupp, I. A. Smellie, *Chem. Eur. J.* **2016**, 22, 15328.
S. E. Ashbrook, G. P. M. Bignami, M. Bühl, D. B. Cordes, D. M. Dawson, R. A. Doyle, Z. Ke, F. M. Mack, A. M. Z. Slawin, I. A. Smellie, *Chem. Commun.* **2017**, 53, 10512

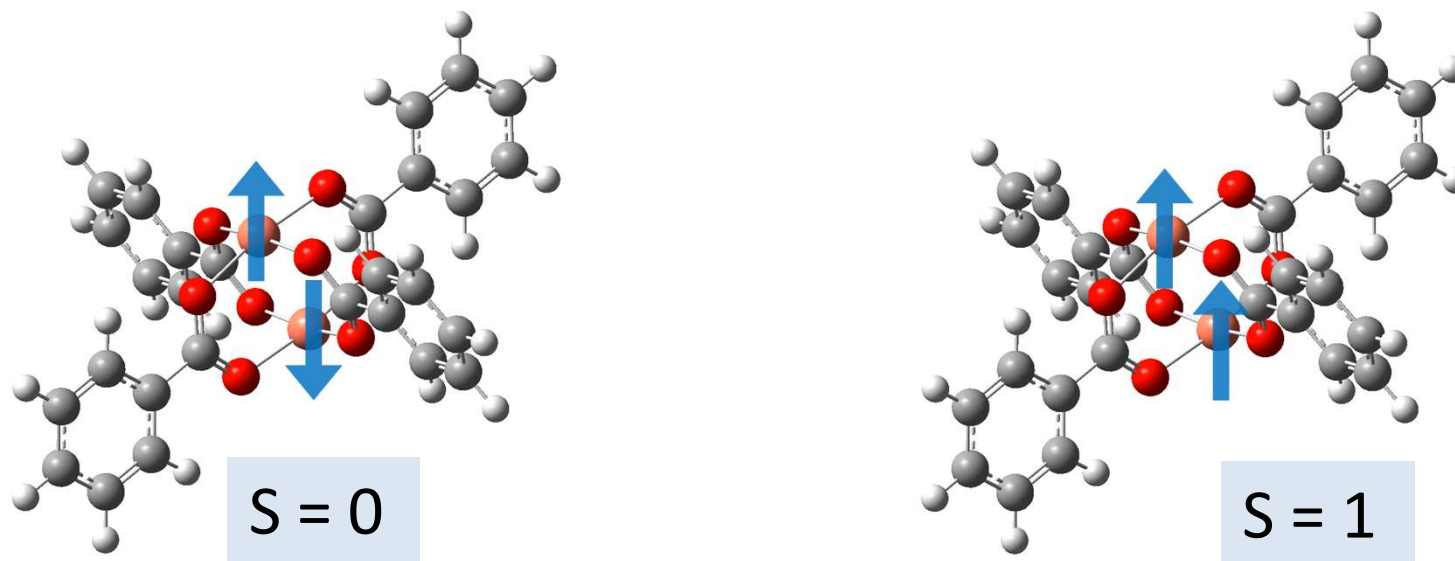
Molecular model for MOFs

Cu(II) benzoate, isolated paddle-wheel dimers



Molecular model for MOFs

Problem: singlet ground state (no pNMR shifts expected)



$$\Delta E_{ST} = 134.6 \text{ cm}^{-1}$$

(PBE0- $\frac{1}{3}$ /IGLO-II//PBE0-D3/AE1)

Molecular model for MOFs

Working hypothesis: pNMR shifts arise from thermal population of triplet state

➔ Boltzmann averaging:

$$x_i = N_i/N_{\text{total}} = g_i \exp(-\Delta E_i/RT) / \sum_i g_i \exp(-\Delta E_i/RT)$$

$$\delta_{\text{total}} = \sum x_i \delta_{\text{iso}(i)}$$

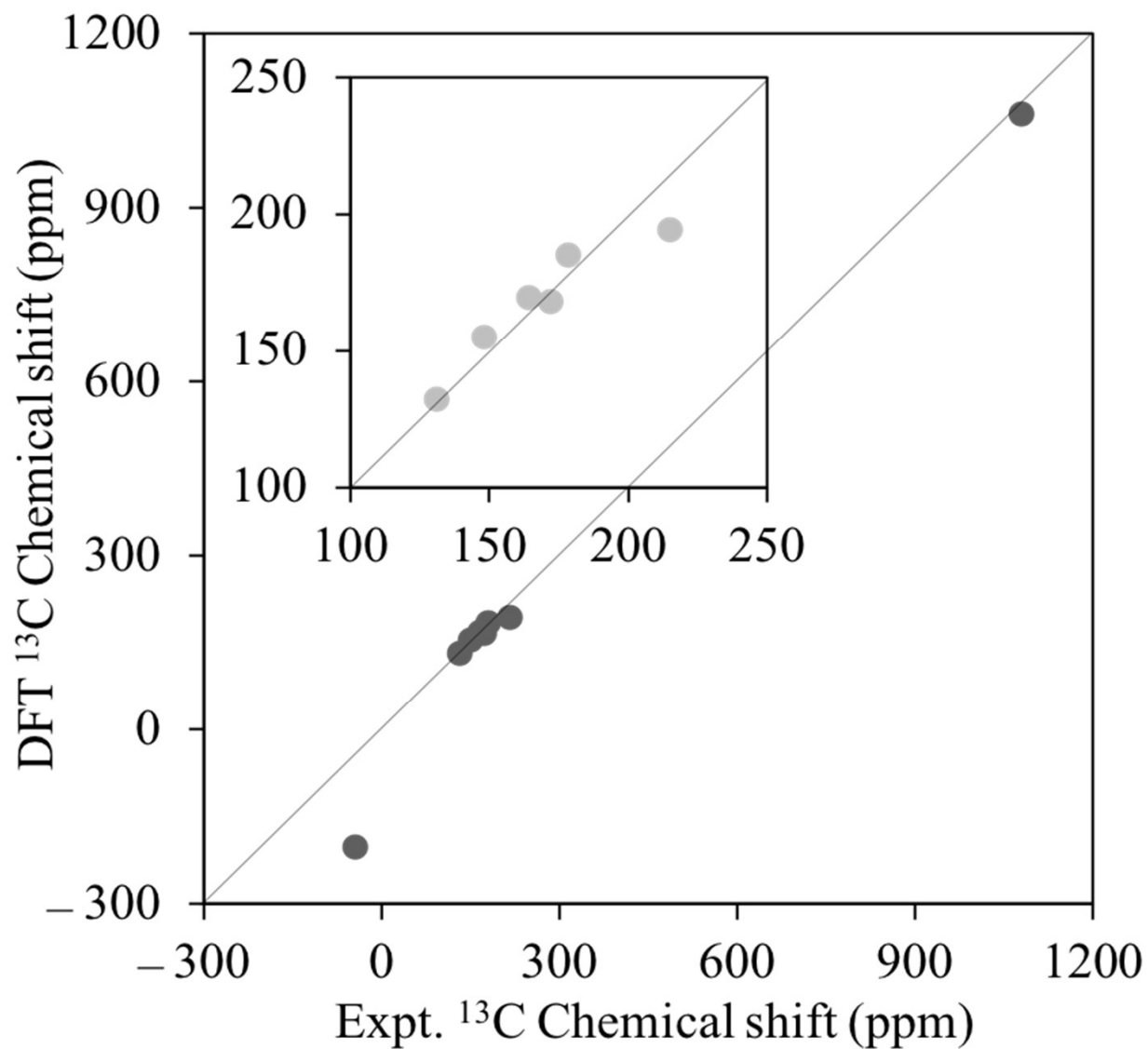
$i = 1(\text{singlet}): \Delta E_1 = 0, \quad g_1 = 1$

$i = 2(\text{triplet}): \Delta E_2 = \Delta E_{\text{ST}}, \quad g_2 = 3$

σ_{orb} calculated using broken-symmetry
Kohn-Sham DFT

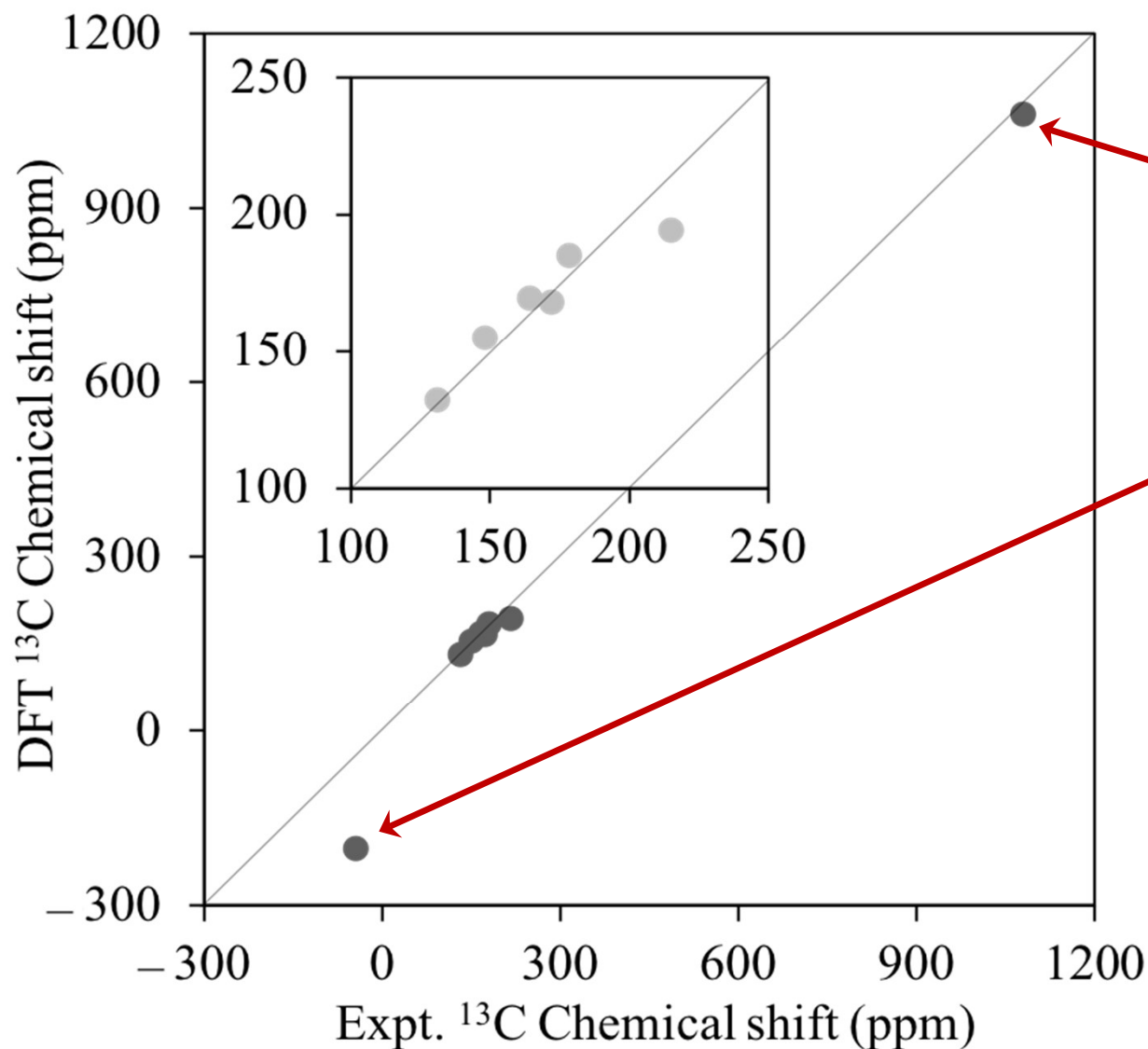
PBE0- $1/3$
/IGLO-II

Performance of the model

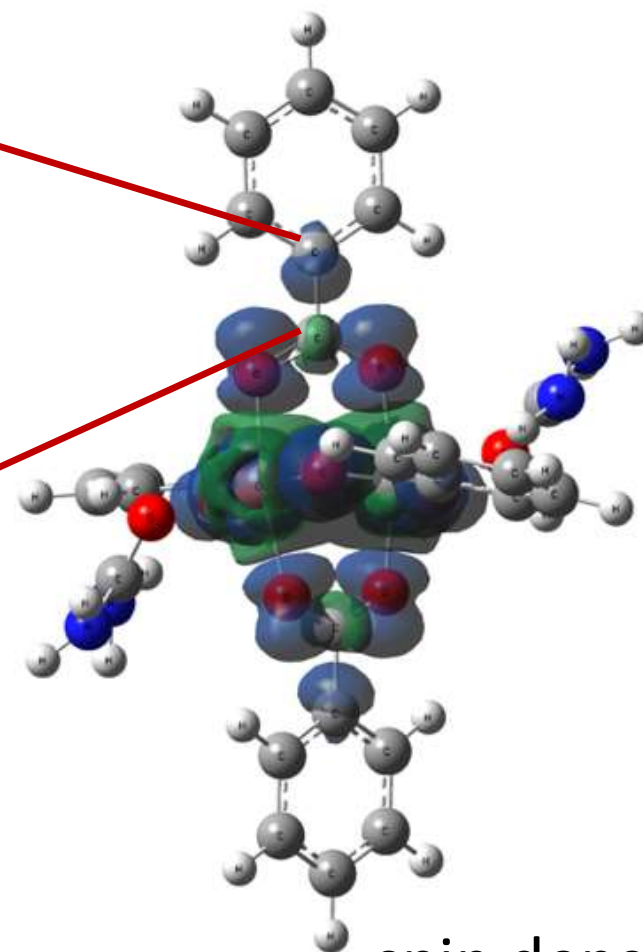


PBE0- $1/3$
/IGLO-II

Performance of the model

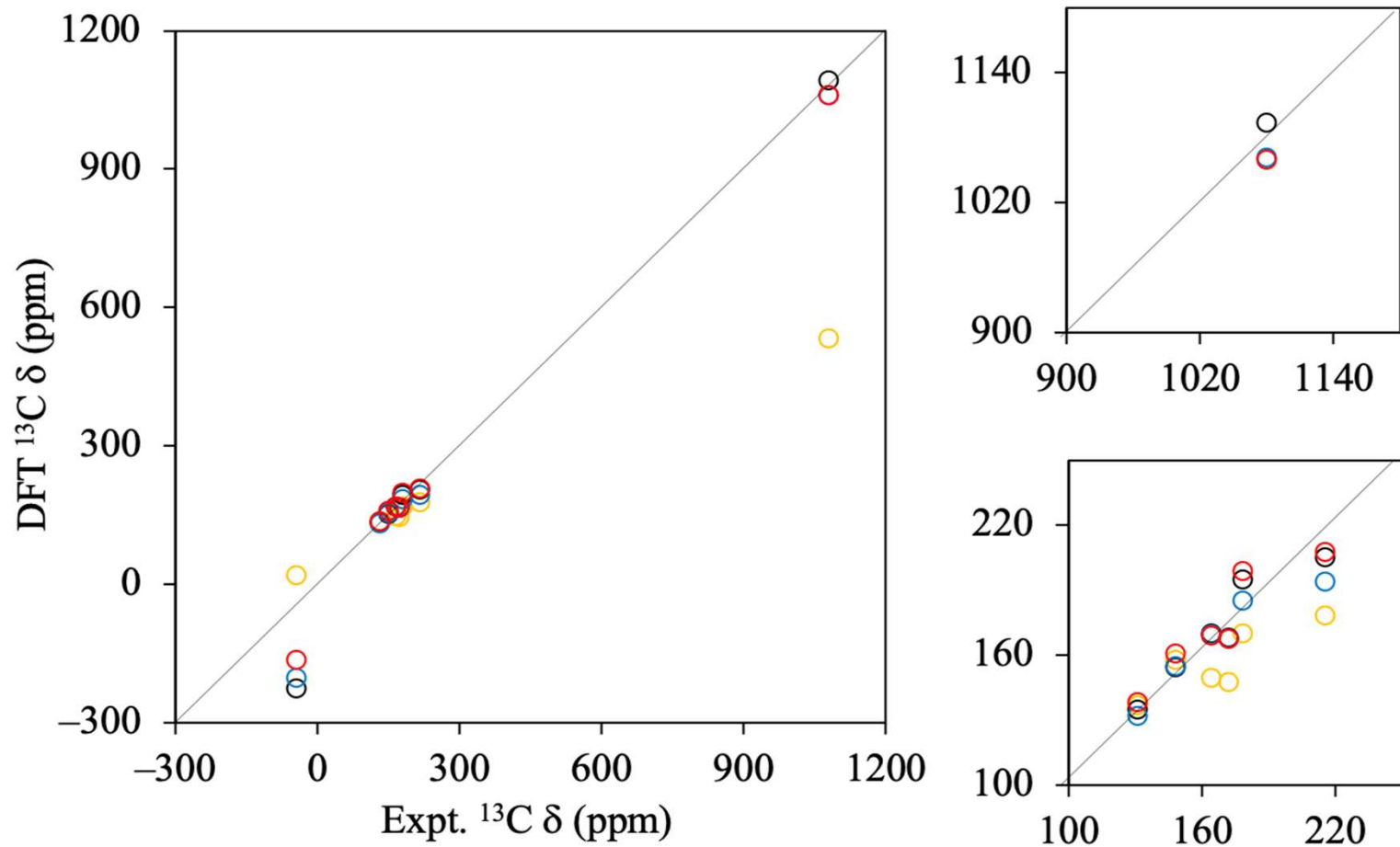


assignment



spin density
(PBE0- $1/3$ /IGLO-II)

Effect of DFT functional

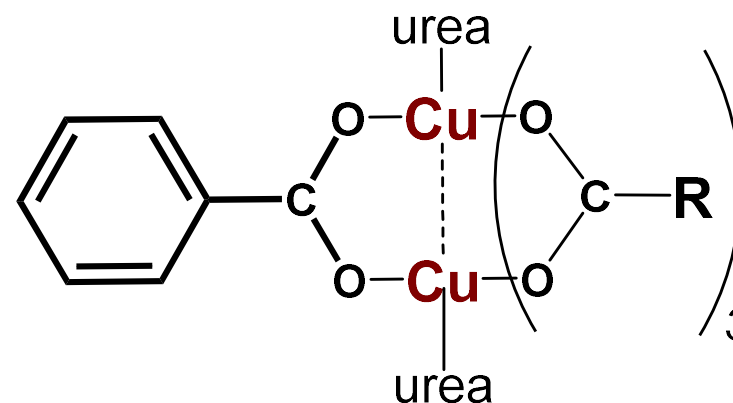


○ PBE ○ PBE0 ○ PBE0- $\frac{1}{3}$ ○ CAM-B3LYP

CAM-B3LYP slightly superior

Effect of structure

Level of pNMR calculations: CAM-B3LYP/IGLO-II



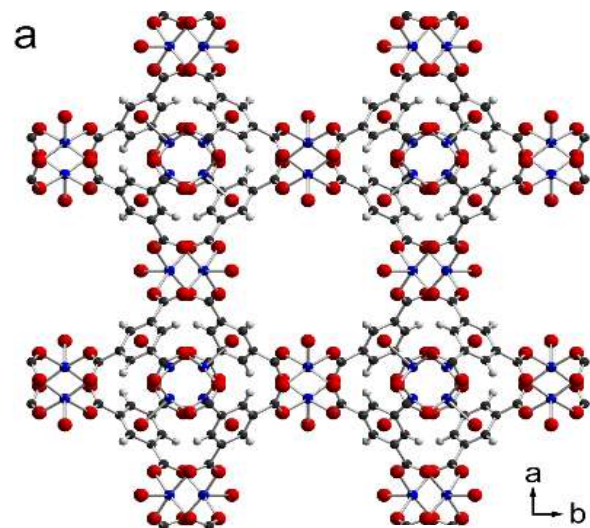
Level of opt	Substituents	MAD
PBE0-D3	R = Ph	23.0
GFN2-xTB ^{a,b}	R = Ph	22.6
GFN2-xTB ^{a,b}	R = Me	21.7

significant computational savings

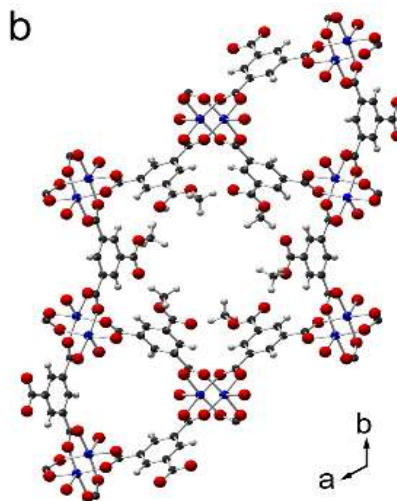
^aC. Bannwarth, S.Ehlert, S.; Grimme, *J. Chem. Theory Comput.* **2019**, *15*, 1652.

^bHigh-spin structure also used for BS singlet.

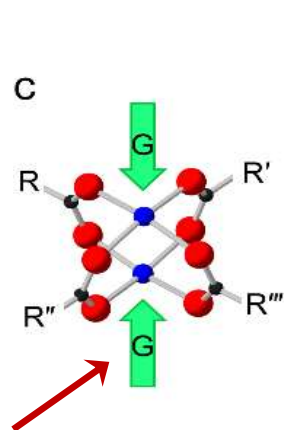
Multinuclear MOF models



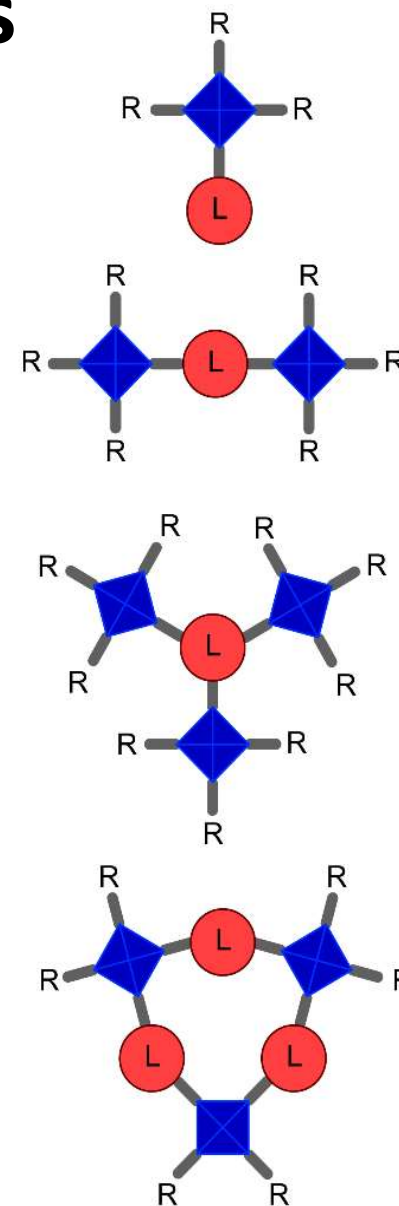
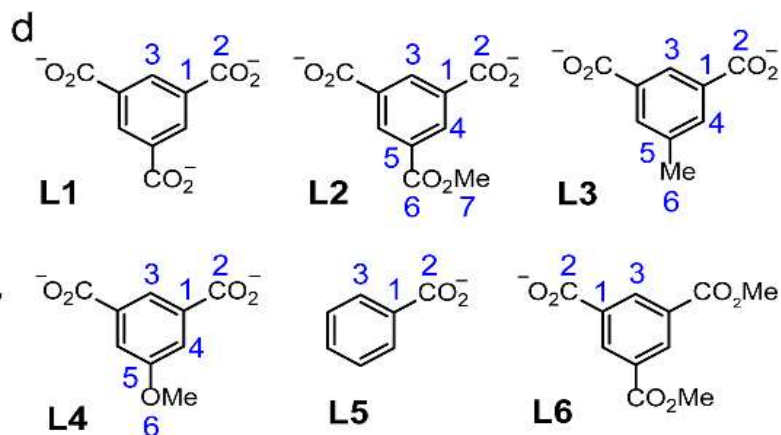
HKUST



STAM-1

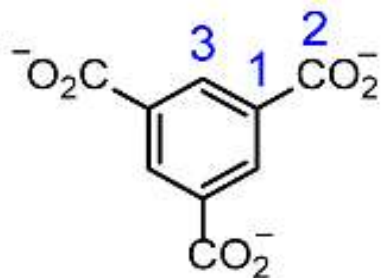


G = H₂O
or none

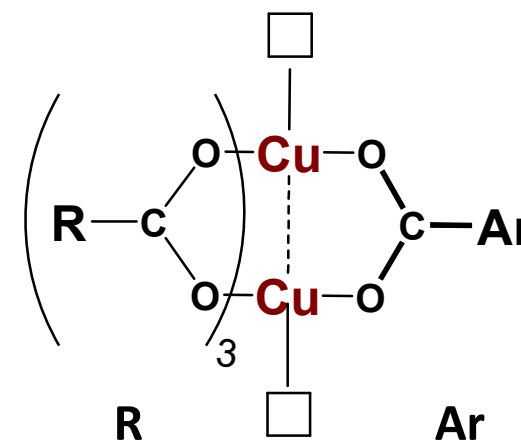


Single dimer model for activated HKUST

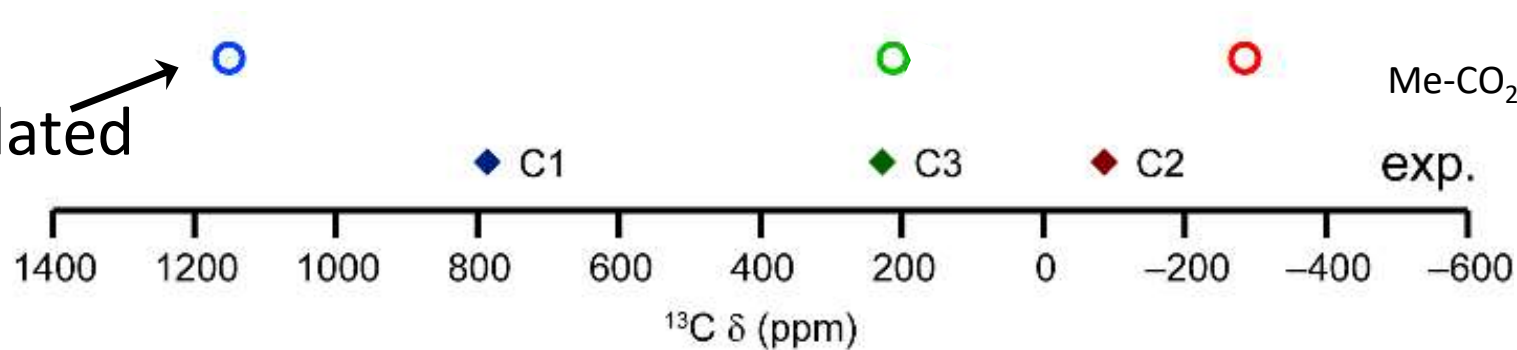
exptl.
linker:



mononuclear
models:

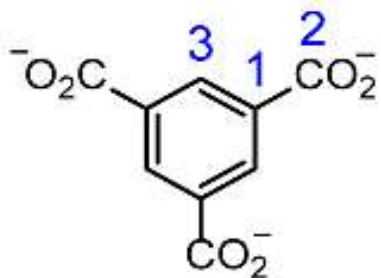


o: as
calculated

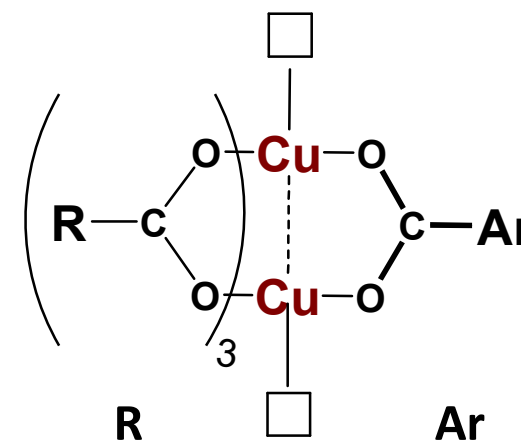


Single dimer model for activated HKUST

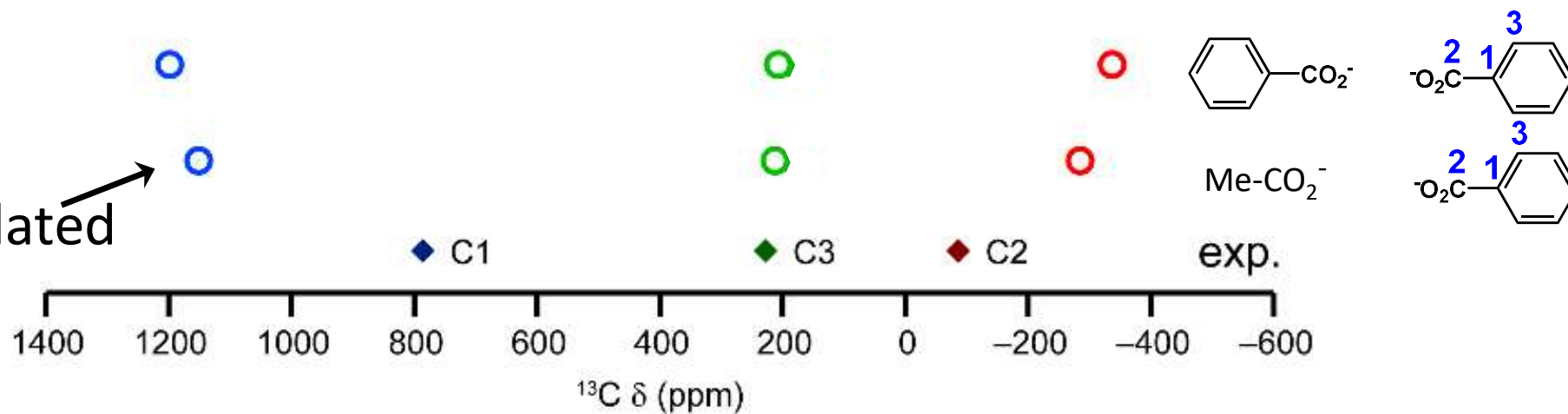
exptl.
linker:



mononuclear
models:

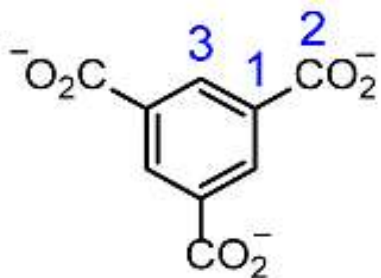


o: as
calculated

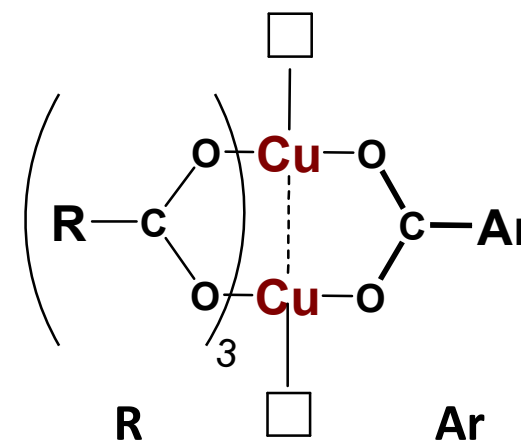


Single dimer model for activated HKUST

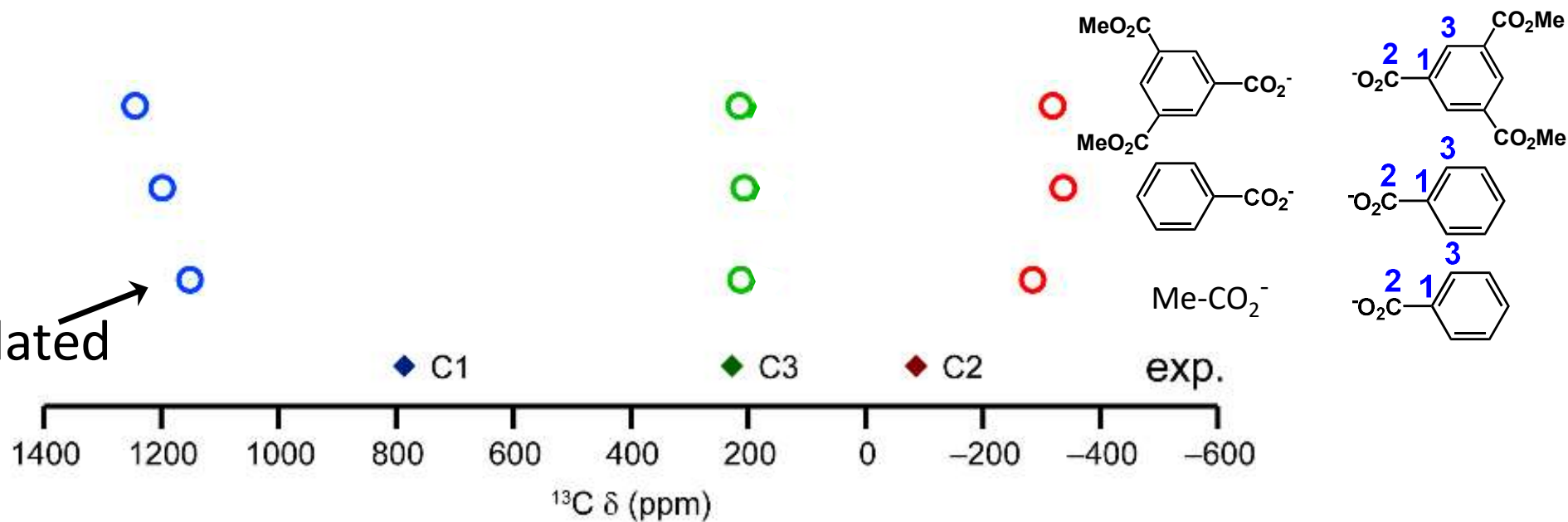
exptl.
linker:



mononuclear
models:



o: as
calculated



Single dimer model for activated HKUST

Pragmatic way to reduce contributions from triplet:

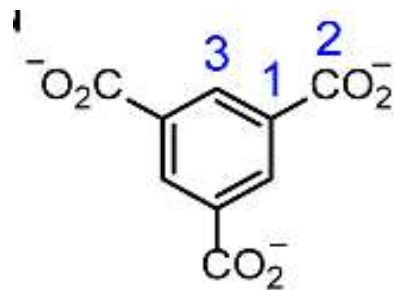
- Scaling of S-T gap (ΔE_{ST}) in Boltzmann distribution

$$x_{\text{trip}} = g_{\text{trip}} \exp(-s\Delta E_{ST}/RT) / [1 + g_{\text{trip}} \exp(-s\Delta E_{ST}/RT)]$$

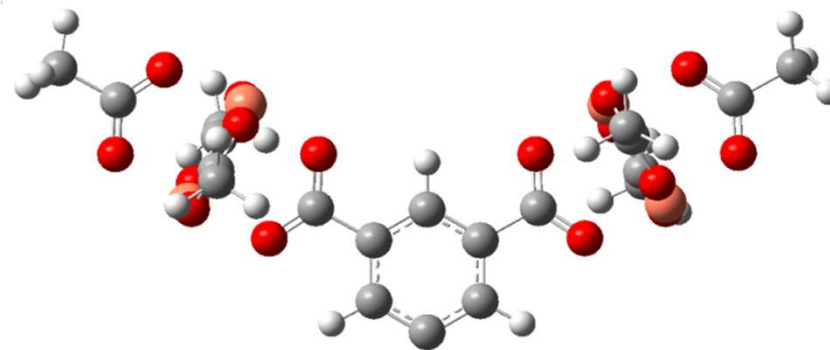
scaling factor **s**

Double dimer model for activated HKUST

exptl.
linker:



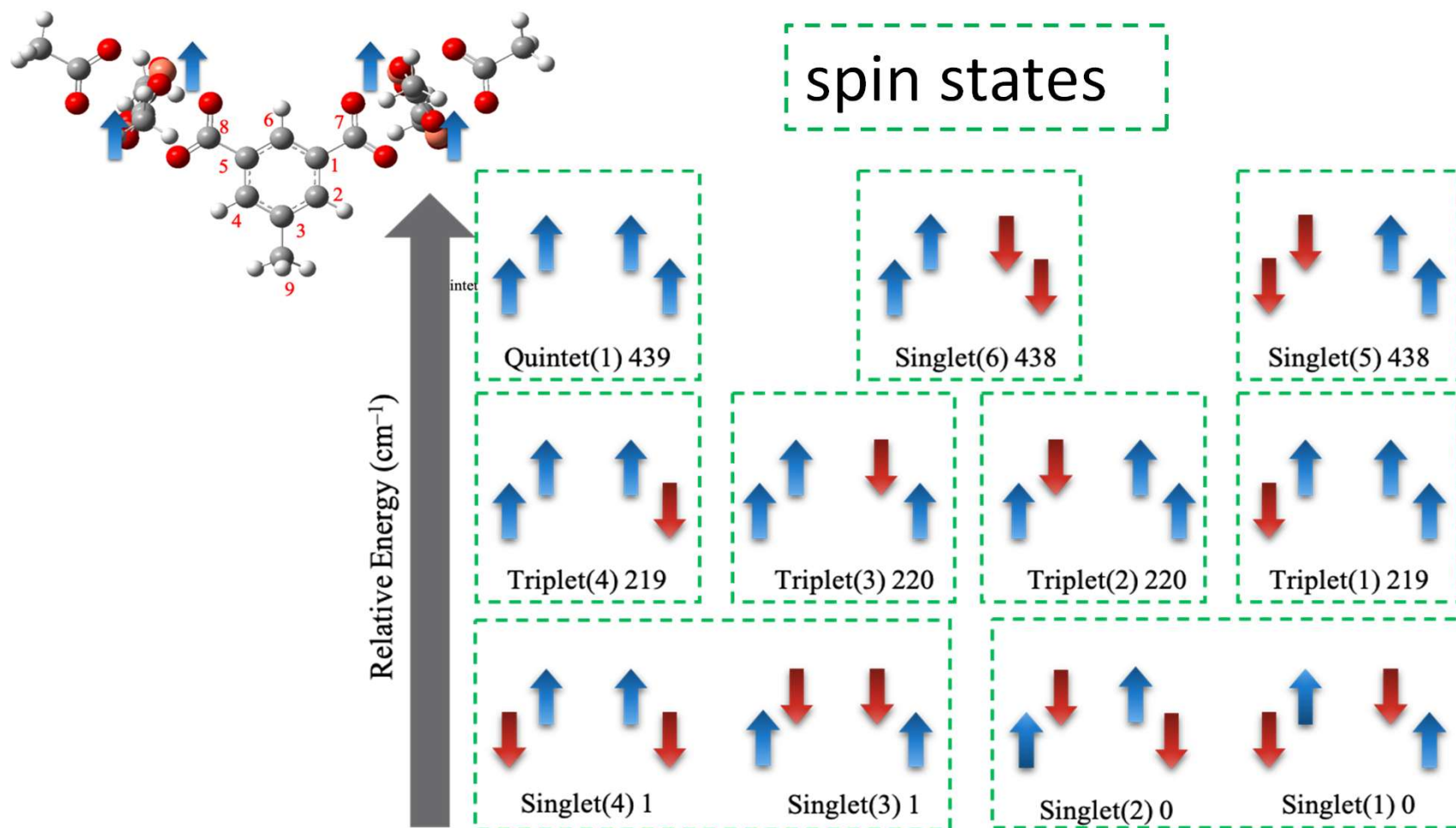
dinuclear models:



Problem:

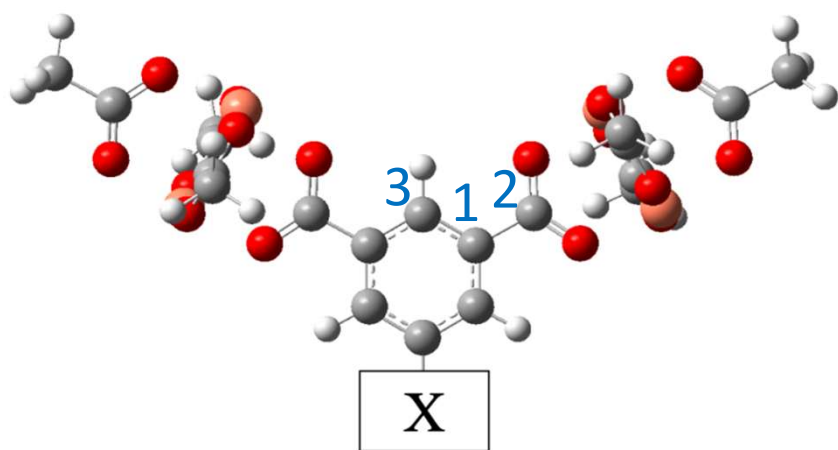
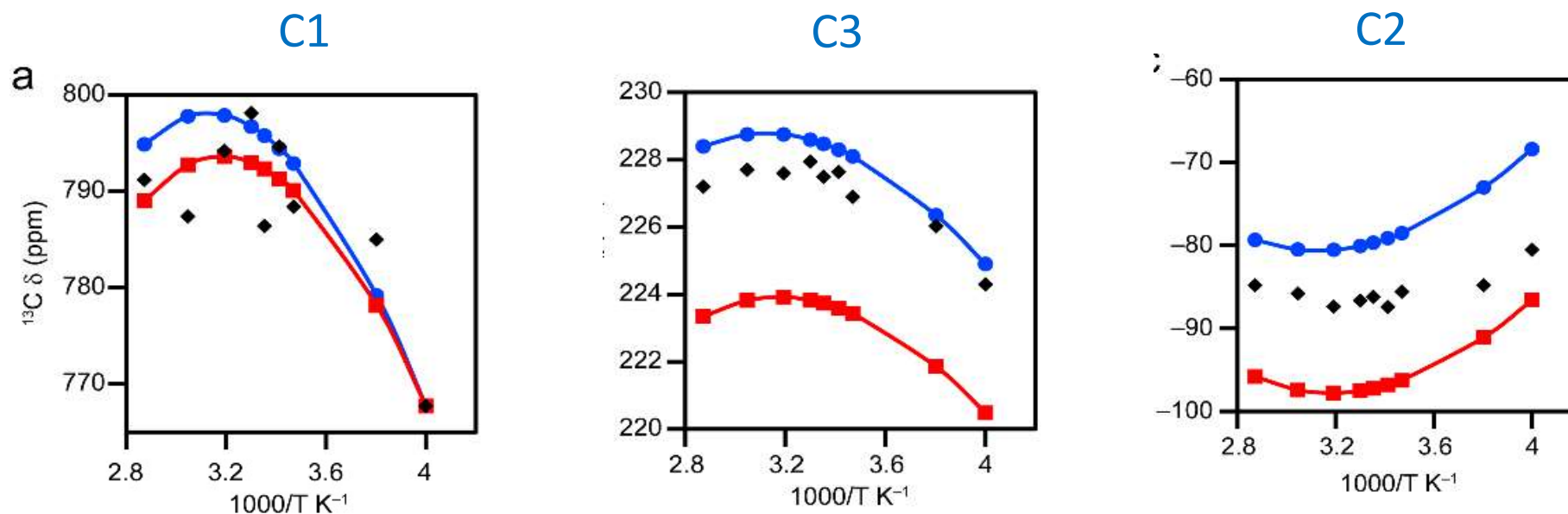
Many possible spin configurations
How to combine into spin states?

Double dimer model for activated HKUST



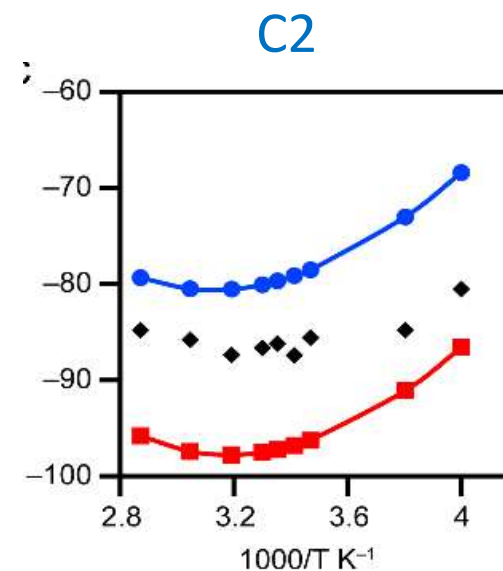
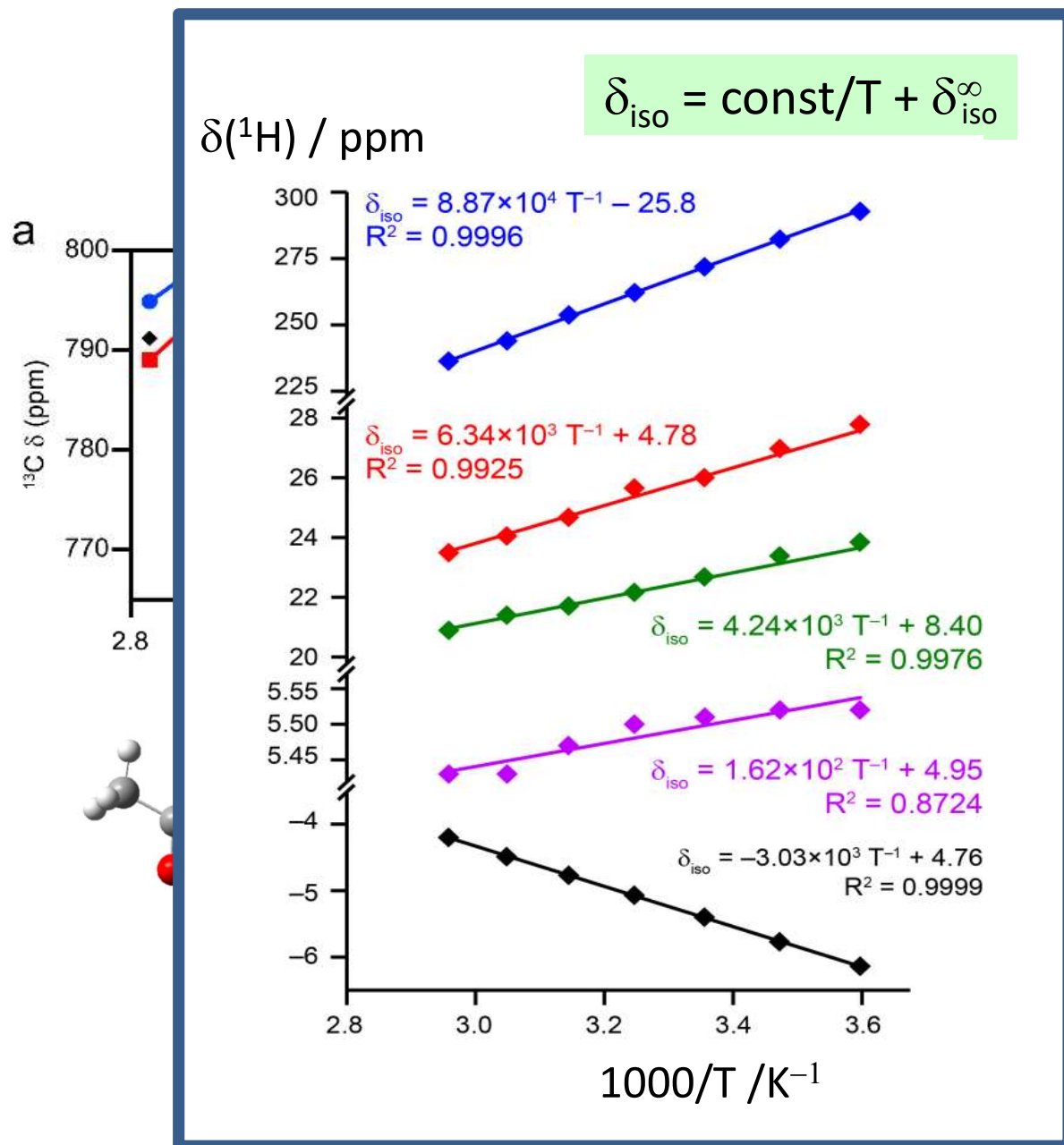
single scale factor: $x_i = g_i \exp(-s\Delta E_i/RT) / \sum_i g_i \exp(-s\Delta E_i/RT)$

Temperature dependence



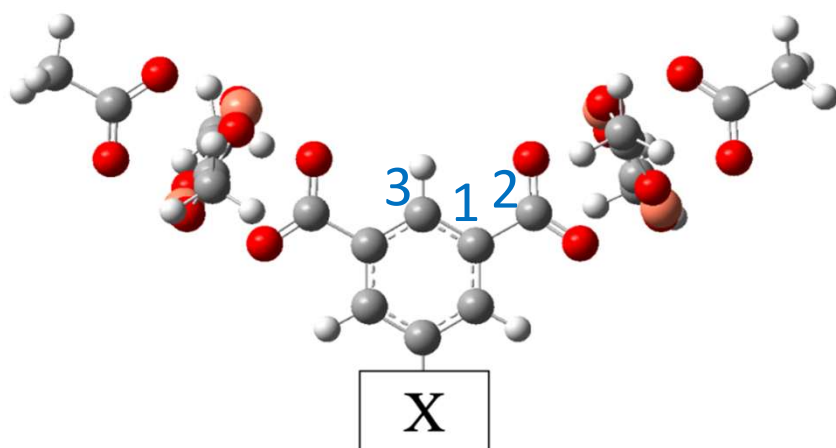
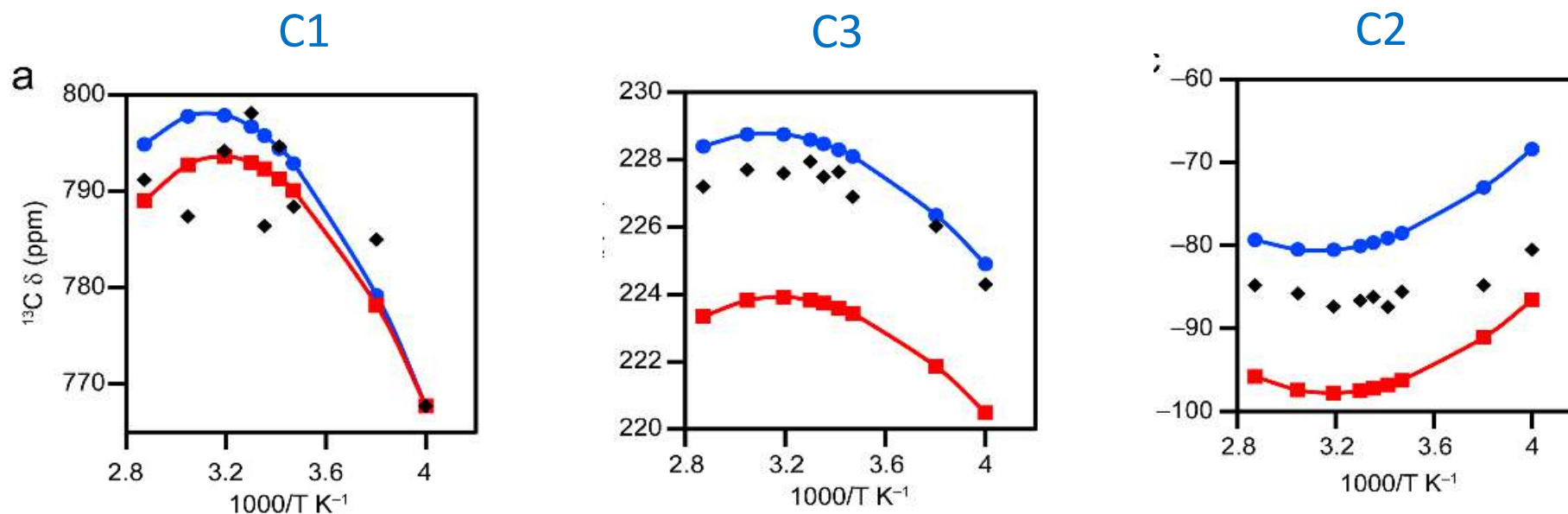
- $X = \text{CO}_2\text{Me}$ ($s = 1.33$)
- $X = \text{Me}$ ($s = 1.31$)
- ◆ experiment

dependence



- $X = \text{CO}_2\text{Me}$ ($s = 1.33$)
- $X = \text{Me}$ ($s = 1.30$)
- ◆ experiment

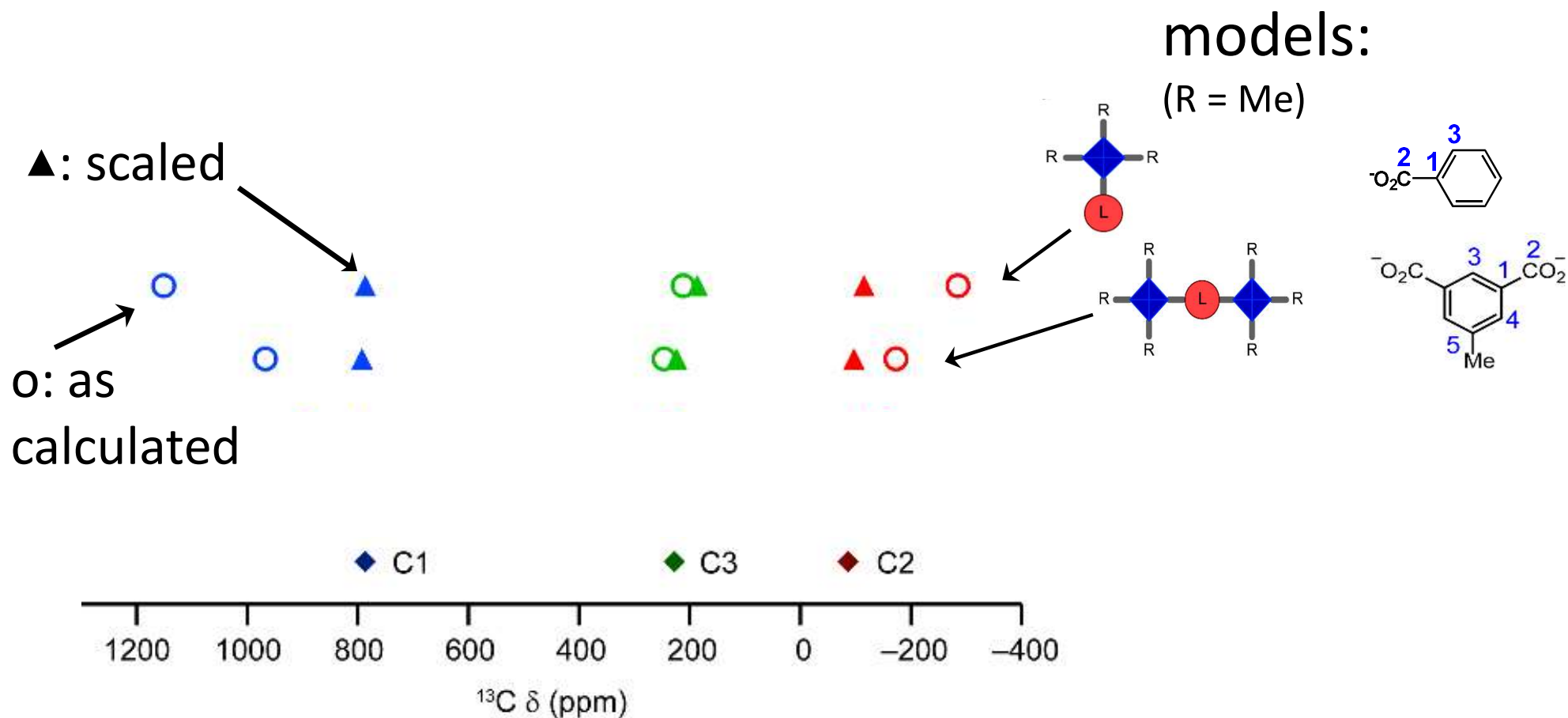
Temperature dependence



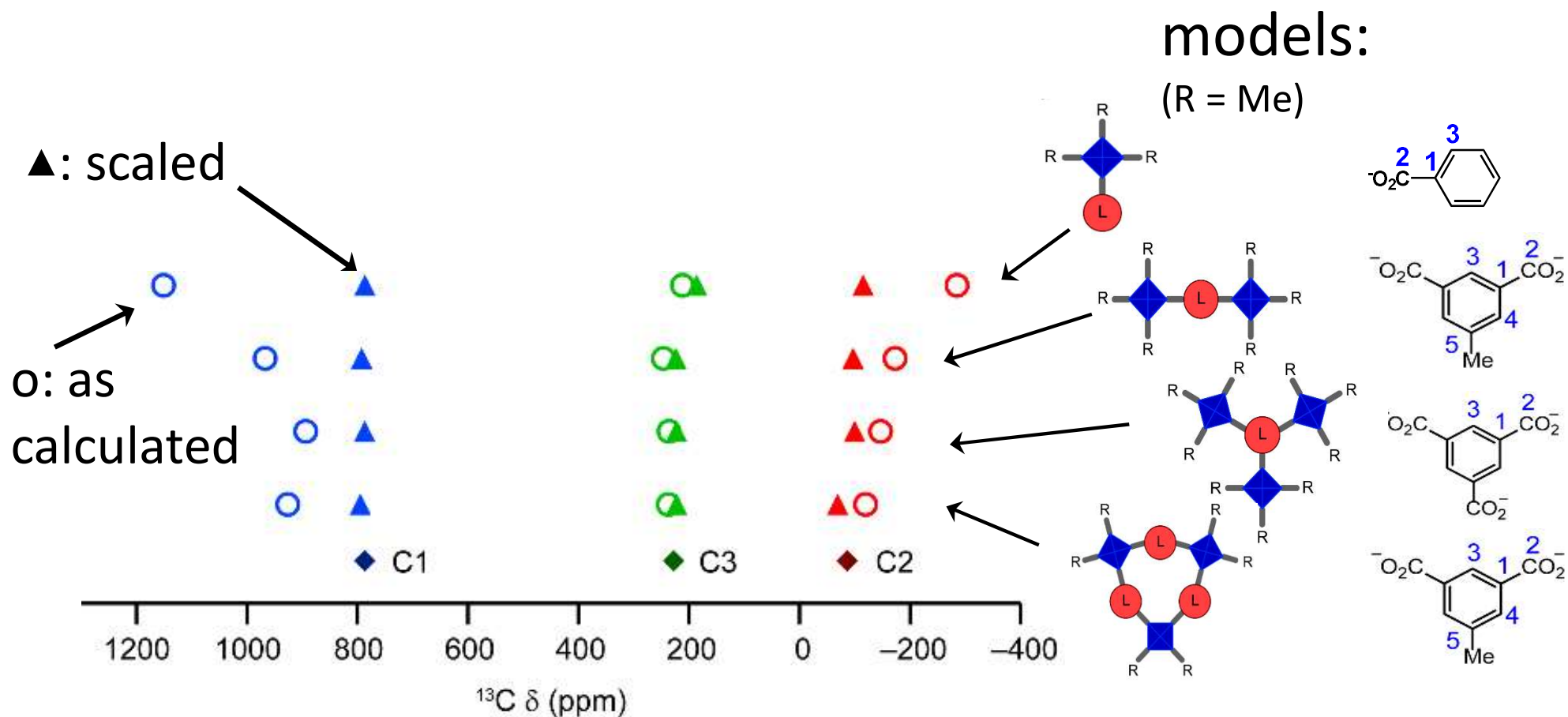
- $X = \text{CO}_2\text{Me}$ ($s = 1.33$)
- $X = \text{Me}$ ($s = 1.31$)
- ◆ experiment

unusual temperature dependence very well reproduced

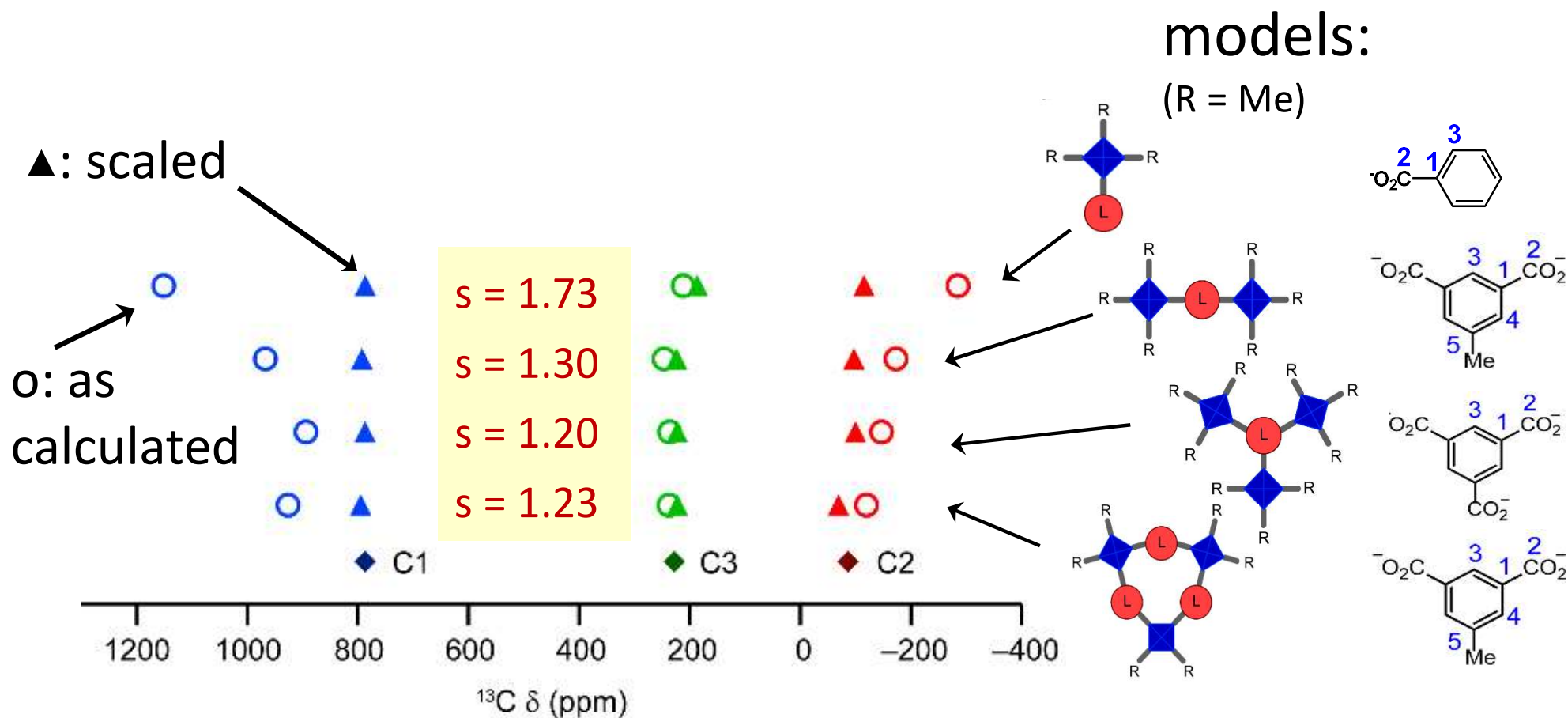
Triple-dimer models for activated HKUST



Triple-dimer models for activated HKUST

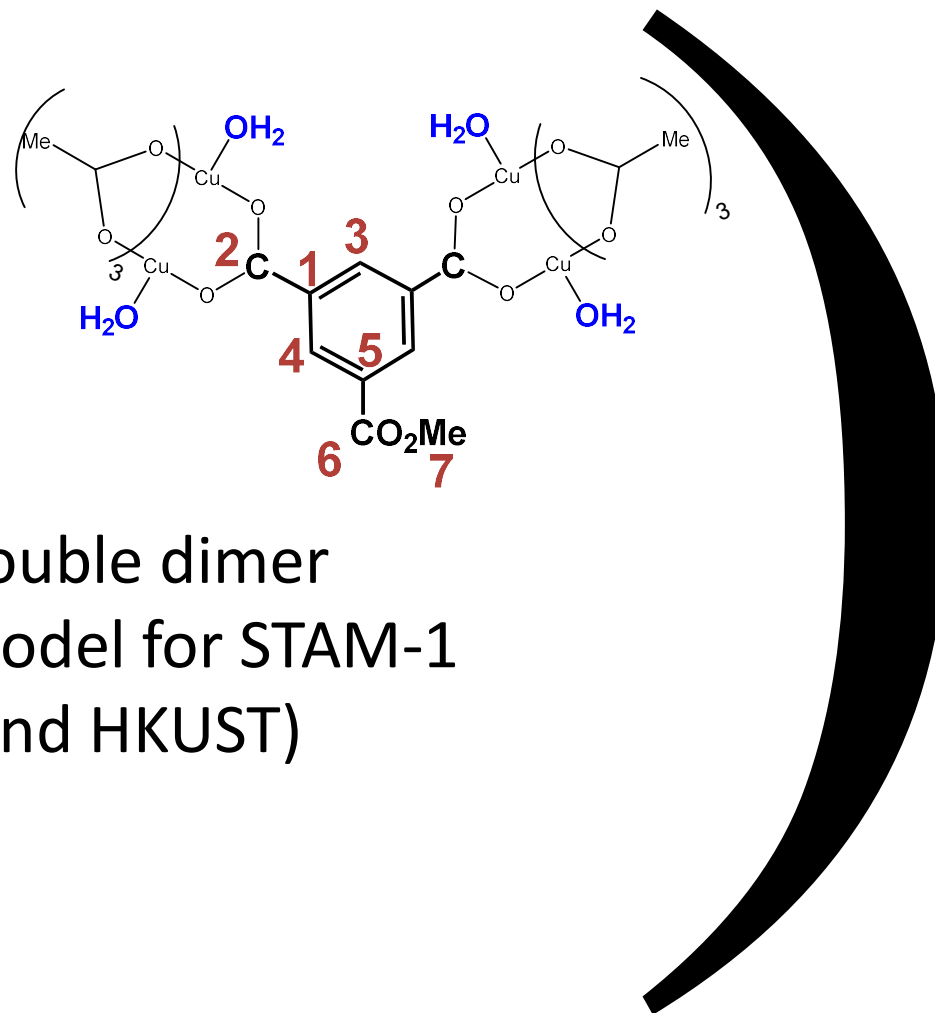
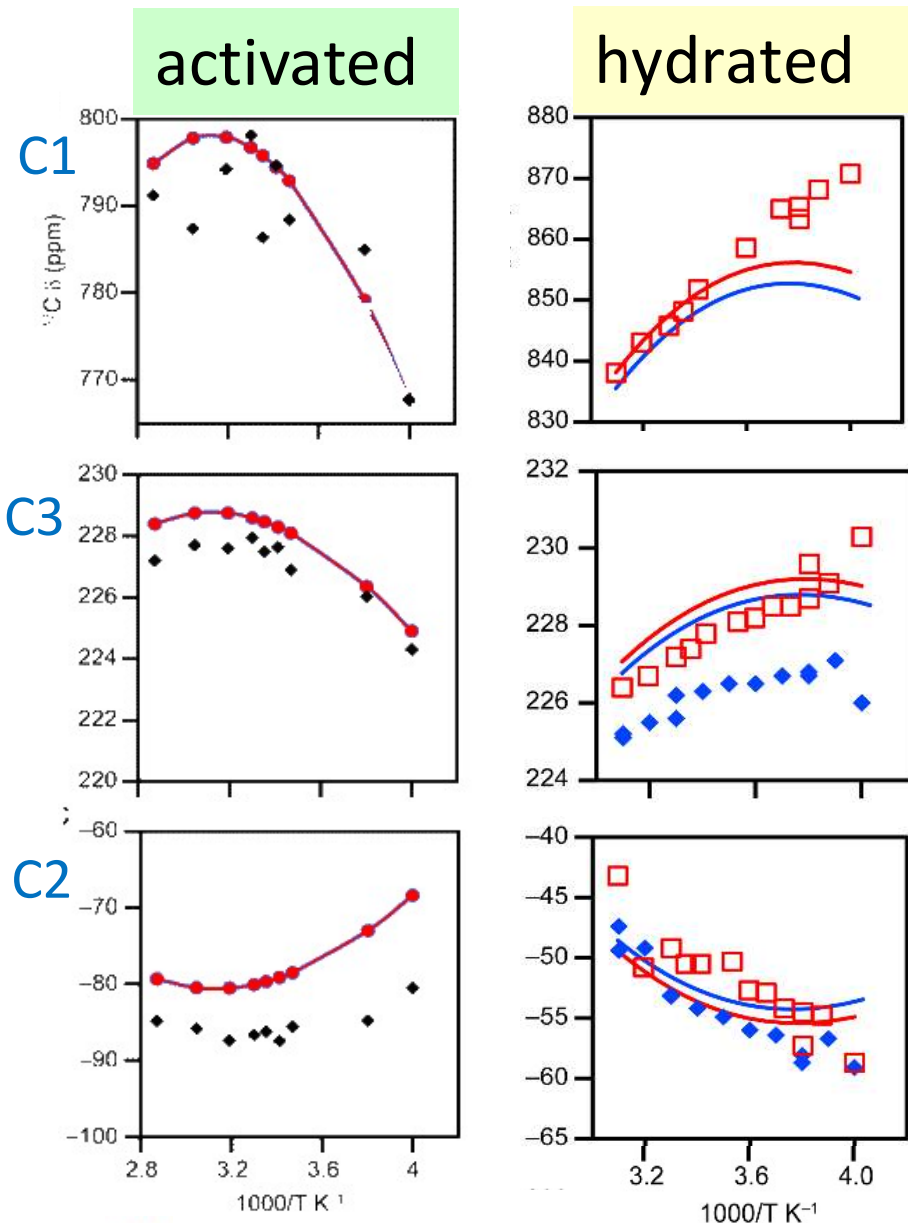


Triple-dimer models for activated HKUST

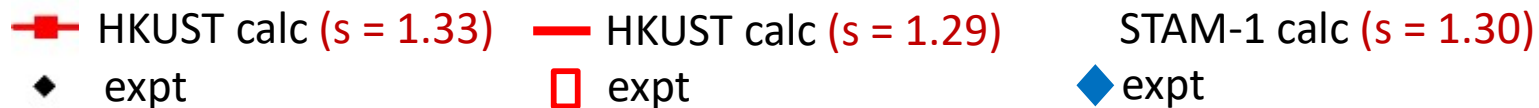


scale factor approaching unity as models become larger

Hydrated MOFs



double dimer
model for STAM-1
(and HKUST)



Conclusion

- Molecular models for MOFs validated in "bottoms-up" approach
- pNMR shifts arise from thermal population of high-spin states
- Uniform scaling of as-calculated energy gaps required
- Temperature dependence of pNMR signals explained

Next challenge:

refinement of models (dynamics, crystal matrix)

Develop pNMR calculations and experiments into structural tool, e.g. to study uptake of guest molecules